

Molecular Dynamics Simulations of Influence of Surface Temperature on Fluorine Etching of Silicon

Jianwei Wang, Yixu Song, Xiaodi Deng, Jinchun Li, and Tianling Ren

Abstract—Molecular dynamics simulation of the reactions between gaseous fluorine atoms and silicon are performed using the development Tersoff-Brenner potential at the temperature from 500K to 1200K. The simulation results show that the Si surface temperature significantly affects the F etching. For instance, as the surface temperature rises up, the numbers of F atoms deposited on and scattered by Si surface decrease, at the same times, the number of the sputtering fluorine atoms and the reactive F atoms with surface to produce volatile compounds increase. In addition, the quantity of the F etched Si atoms increased with an increase of the surface temperature.

Index Terms—Molecular dynamics, etching, temperature, silicon, fluorine.

I. INTRODUCTION

In recent years, plasma etching technology is widely used in the fields of microelectronics and semiconductor [1]-[3]. Over the years, people have been using halogen group (F, Cl, Br) plasma silicon material etching^[4-7]. Searching Fluorine plasma etching mechanism of Si for improving the etching process to improve the etching efficiency, has a great significance. Due to the complexity of the plasma environment as well as its complexity and the role of the material surface and experimental conditions limitations, which makes it difficult to fully understand the etching microscopic mechanism.

As a way to research Micro-physical properties of material, molecular dynamics(MD) simulation method has been widely used to study plasma surface interaction areas, and has made a consistent and experimental results [8]-[12]. In order to study the microscopic mechanism of the fluorine-containing plasma etching of silicon, Graves and Abrams develop the Tersoff-Brenner potential function, the description of this potential function is used to C-Si-F system.

In this paper, MD method has been used to simulate the dynamic process of low energy F atoms etching Si at different temperatures. Analyzes the impact of temperature

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on the etching. The work on the treatment of the F plasma etching of Si has a certain reference value.

II. MODELING AND SIMULATION METHODS

A. Simulation Method

Theoretical work has focused on the use of ab initio calculations and molecular dynamics and molecular dynamics simulation to study reactions between both F atom and F₂ molecular with the Si{100} surface. A potential energy function with two-body and three-body terms to model silicon-silicon, fluorine-fluorine, and silicon-fluorine interactions was first developed by Stillinger and Weber(SW). The three-body terms are necessary to model the tetrahedral bonding of silicon. The limitation of the SW parameterization of the SW potential energy function is that it was fit to experimental gas-phase data on SiF_x Species and, thus, was not designed to model interactions of fluorine with a silicon surface {100} surface, Weakliem, Wu, and Carter (WWC) reparameterized the SW potential [13] for the silicon-fluorine interactions. Abrams and Graves have developed a C-Si-F system potential and based on the Tersoff-Brenner(TB) form potential and used it to investigate CF_x (x=1-3) etching silicon. They also demonstrated that a Si_xC_yF_z reaction layer was formed and its thickness increased with the incident energy. In recent years, the SW and TB potential energy function are widely used. Therefore, in this work, the simulation of silicon-fluorine etching with the Tersoff-Brenner potential which developed by Graves^[14] are performed. Incident fluorine atoms are used with kinetic energies ranging from 2.0eV to 15eV. The reaction of the temperature is from 500K to 1200K. The surface that Si target substrate consists of maximum 1728 atoms and its maximum dimension is 32.568Å×32.568Å×32.568Å, and timestep is 0.1fs. The F atoms injection direction is $\langle 0\bar{1}0 \rangle$.

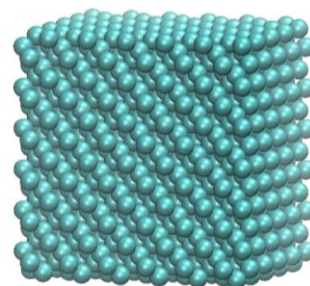


Fig. 1. The silicon model of the MD simulation

B. Si-F System Potential

In the system of the simulation, first of all, we need catch the parameters of the potential. For two elements system,

Si-Si, F-F, and Si-F potential must be considered.

We used the same function forms as Tersoff-Brenner potential. We briefly describe the functional form of the potential here. We describe each atom in a closed system by indices i, j , and k . The interatomic potential energy between two atoms i and j is defined as

$$E_{ij} = V_R(r_{ij}) - B_{ij} V_A(r_{ij}) \quad (1)$$

where r_{ij} is the distance between atoms i and j , V_R and V_A are the repulsive and attractive potentials, and B_{ij} is an average empirical bond-order function. V_R and V_A are defined as

$$V_R(r_{ij}) = f_{ij}(r_{ij}) \frac{D_{ij}^{(e)}}{S_{ij} - 1} \exp[-\sqrt{2S_{ij}} \beta_{ij} (r_{ij} - R_{ij}^{(e)})] \quad (2)$$

and

$$V_A(r_{ij}) = f_{ij}(r_{ij}) \frac{D_{ij}^{(e)} S_{ij}}{S_{ij} - 1} \exp[-\sqrt{2S_{ij}} \beta_{ij} (r_{ij} - R_{ij}^{(e)})] \quad (3)$$

Here S_{ij} is a constant, and $R_{ij}^{(e)}$ and $D_{ij}^{(e)}$ are equilibrium bond length and bond energy when the bond order is one. Exponent constant β_{ij} is related to the force constant of the bond as

$$\beta_{ij} = \sqrt{k / (2D_{ij}^{(e)})} \quad (4)$$

In order to cut off the potential smoothly, a switching function f_{ij} is defined as follows.

$$f_{ij}(r_{ij}) = \begin{cases} 1 & r < R_{ij}^{(1)} \\ \{1 + \cos[\pi(r_{ij} - R_{ij}^{(1)}) / (R_{ij}^{(2)} - R_{ij}^{(1)})]\} & R_{ij}^{(1)} < r < R_{ij}^{(2)} \\ 0 & r > R_{ij}^{(2)} \end{cases} \quad (5)$$

The total number of atoms bonded to atom i , $N_i^{(t)}$, the number of adjoining fluorine atoms $N_i^{(F)}$ is important to determine bond orders.

$$N_i^{(F)} = \sum_{k \in F, k \neq i, j} f_{ik}(r_{ik}) \quad (6)$$

The details of the potential can refer to Ref. 14.

C. After-Treatment

We use the open source code LAMMPS as the main program to run the MD simulation calculation. The particle incident is continuous. To a single particle the interaction time is 1ps, and the timestep is 0.1fs. When binding energy between a particle and surface of the pattern is lower the threshold energy, the particle can be removed as a product. The threshold energy can expressed as follows.

$$E_b = k_B T \ln(\tau A) \quad (6)$$

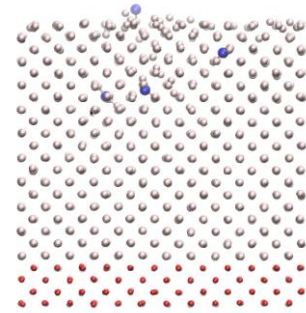
T is the surface temperature and τ is the time constant, A is a constant. In this simulation the value of A and τ is 10^{12} s^{-1}

and $1 \mu\text{s}$. k_B is Boltzmann constant.

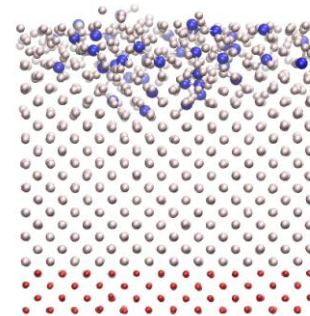
III. RESULTS AND DISCUSSION

In the simulation process of fluorine atoms etching the surface of silicon, verlet algorithm is used to solve Newton's equation of motion. The initial position of the incident F particles randomly set in the xy direction, maintained in a horizontal z -direction in a same altitude. The injecting direction of F particles is need to be kept perpendicular to the ground. In the samples of x and y direction using periodic boundary condition and the bottom four layers atoms position need fixed on.

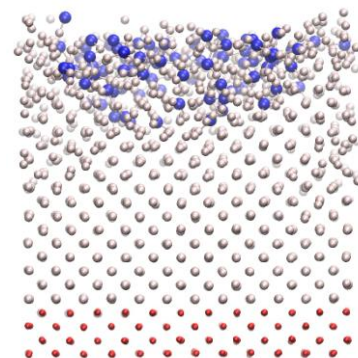
Fig. 2 illustrates at different times the snapshots of the simulation. From the picture we find that the more the time went on, the more the F atoms injected into the Si surface. And at the same time, in the surface of silicon, the F atoms that injected into the Si formed a reaction layer. The thickness of the reaction layer increase over time to reach a peak. At first F atoms and Si atoms formed SiF, than the SiF and another F atom formed SiF₂, SiF₃, and SiF₄. The SiF₄ as a gas product to be removed. Then the silicon surface was etched by F atoms. As the F atoms inject constantly, the surface etched at the same time.



(a)



(b)



(c)

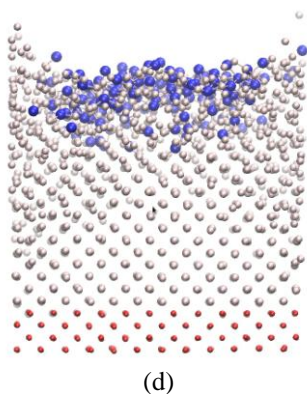


Fig. 2. The snapshots of the simulation at different times(1ps, 50ps, 100ps,200ps).The white balls are the silicon atoms, the blue ones are the F atoms. The temperature is 300K and the incident atom energy is 10eV.

Fig. 3 illustrates at different temperatures for different energy bombardment of F atoms is deposited on the surface of the sample after 500 times. From the figure we could find that the amount of deposition of F atoms, In the initial stage, increasing dramatically with the rise of time. The deposited amount of F atoms on the surface has a direct relationship with temperature of the sample surface. The deposition amount of F atoms is reducing with rise of temperature. Comparison for the 0.5 eV and 10eV, we found that the deposited amount of F atoms increases with energy increasing. Deposited on the surface of the sample, scattering and after deposition of the sputtering may occurred later in the injected F atom, and the sample surface atomic. Only less than 10 percent injected F atoms will deposit on the surface of silicon.

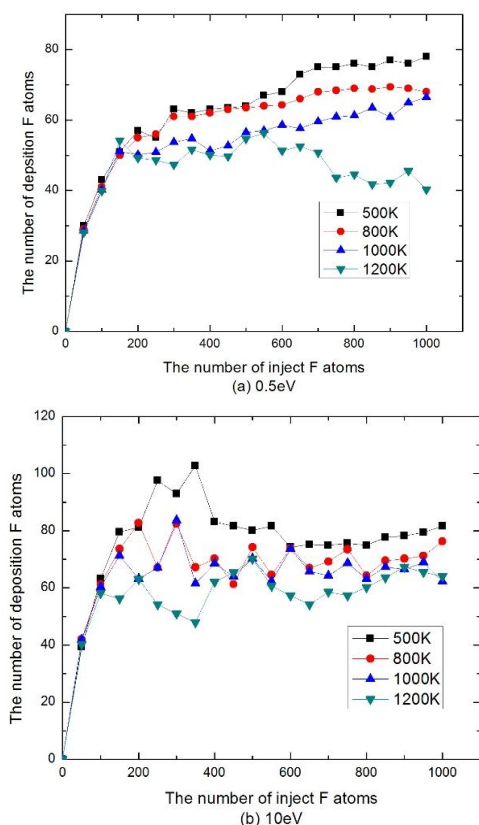


Fig. 3. The number of deposition F atoms as a function of the number of inject F atoms.

Fig. 4 shows when the etching yield of Si atoms in the sample of Si, as a function of the number of injected F atoms at different temperatures. From the picture we can find that in the two energy condition, the etching yield of Si is increasing with the rise of temperature. The Fig. 4 (a) shows that when the the injected F Atomic energy is 0.5eV and the temperature is 500K or 800K, etching product does not appeared. But when the temperature is beyonged 1000K the etching obviously ocurred. The Fig. 4 (b) showed that if energy of F is 10eV, F atoms etching occurred when the temperature is 500K or beyonged it.

From the two picture we also find that if the incident energy is identical. The etch rate (ratio of the number of etched silicon to the number of incident fluorine atoms) is as high as temperature. And when the temperature is same, the more the energy the more the etch rate.

This result has been confirmed in the experiment [15].

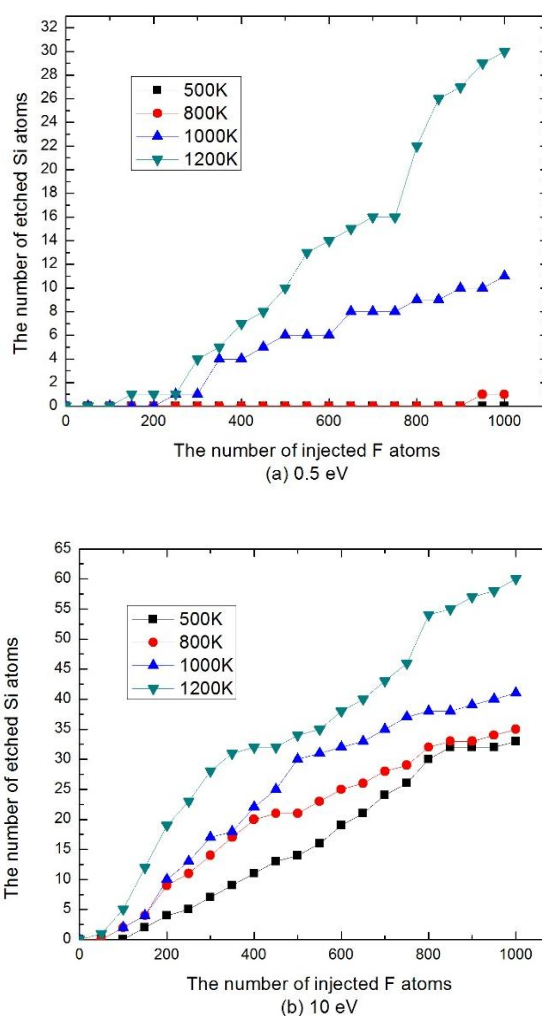


Fig. 4. The number of etching yield F atoms as a function of the number of inject F atoms.

IV. CONCLUSION

For micro-scale MD simulations of fluorocarbon systems, as the surface temperature rises up, the numbers of F atoms deposited on and scattered by Si surface decrease, at the same times, the number of the sputtering fluorine atoms and the reactive F atoms with surface to produce volatile compounds

increase. In addition, the quantity of the F etched Si atoms increased with an increase of the surface temperature.

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